

Nuclear Physics

23



Nuclear Structure Δ :

nuclear properties

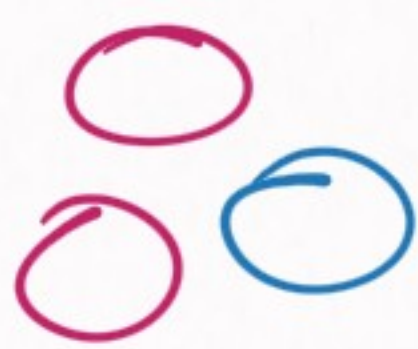
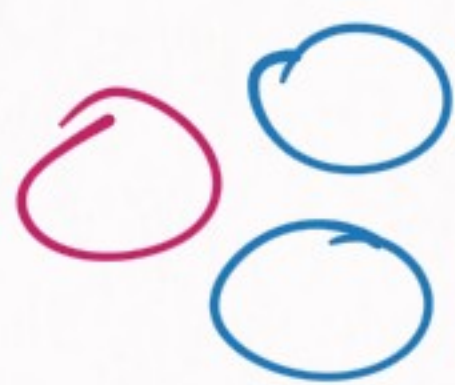
(non-electromagnetic)

Nuclear Structure ? WHY?



Deuteron ($A=2$)

Doable

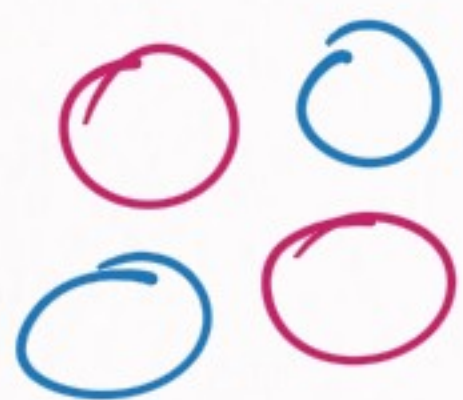


Triton &

Helium-3 ($A=3$)

Still doable

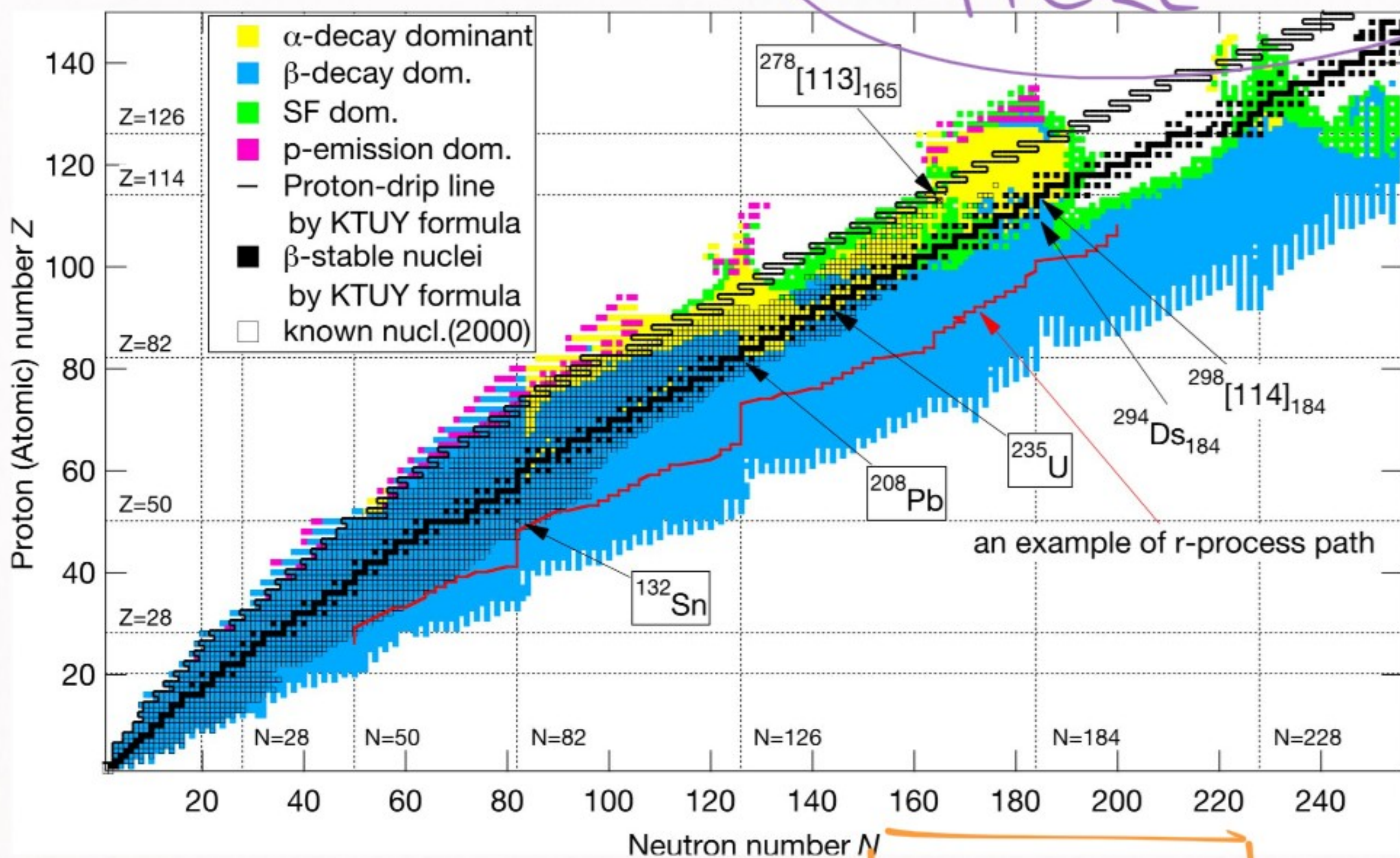
(FADDEEV)



Helium-4

Still...

WHAT ABOUT THESE GUYS HERE



Oups!

A small \Rightarrow Exact methods

A big \Rightarrow We have to think about the structure of a nucleus



SHOPPING LIST

GENERAL
PROPERTIES
OF NUCLEI

What to describe?

1) Binding energy

2) Size

3) Angular momentum &
Parity

4) Magnetic dipole moment

5) Stability / Decays

Nuclear Properties :

1) Binding Energy

$$B_d = (m_p + m_n) - m_d \approx 2.2 \text{ MeV}$$

$$B_t = (m_p + 2m_n) - m_t \approx 8.5 \text{ MeV}$$

↓ We can extend this

$$B(Z, N) = (Zm_p + Nm_n) - m(Z, N)$$

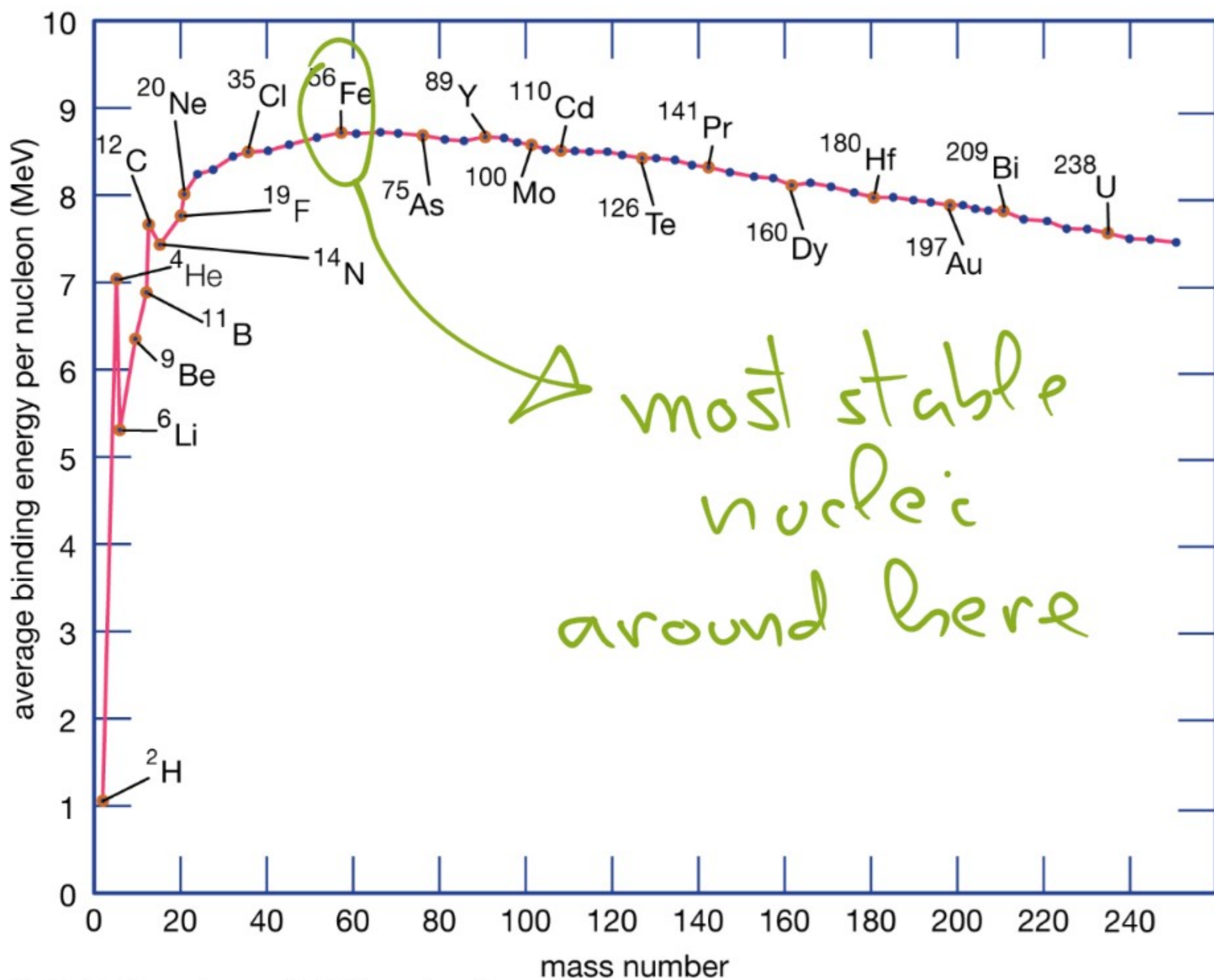
$Z \rightarrow \#$ of protons

$N \rightarrow \#$ of neutrons

$A = N + Z \rightarrow \#$ of nucleons

We find $\frac{B}{A} \approx 8 \text{ MeV/nucleon}$
(SATURATION)

Here's a graphic of $\frac{B}{A}$



© 2012 Encyclopædia Britannica, Inc.

Interesting observation:

$A < 56 \rightarrow$ Fusion possible

$A > 56 \rightarrow$ Fission possible

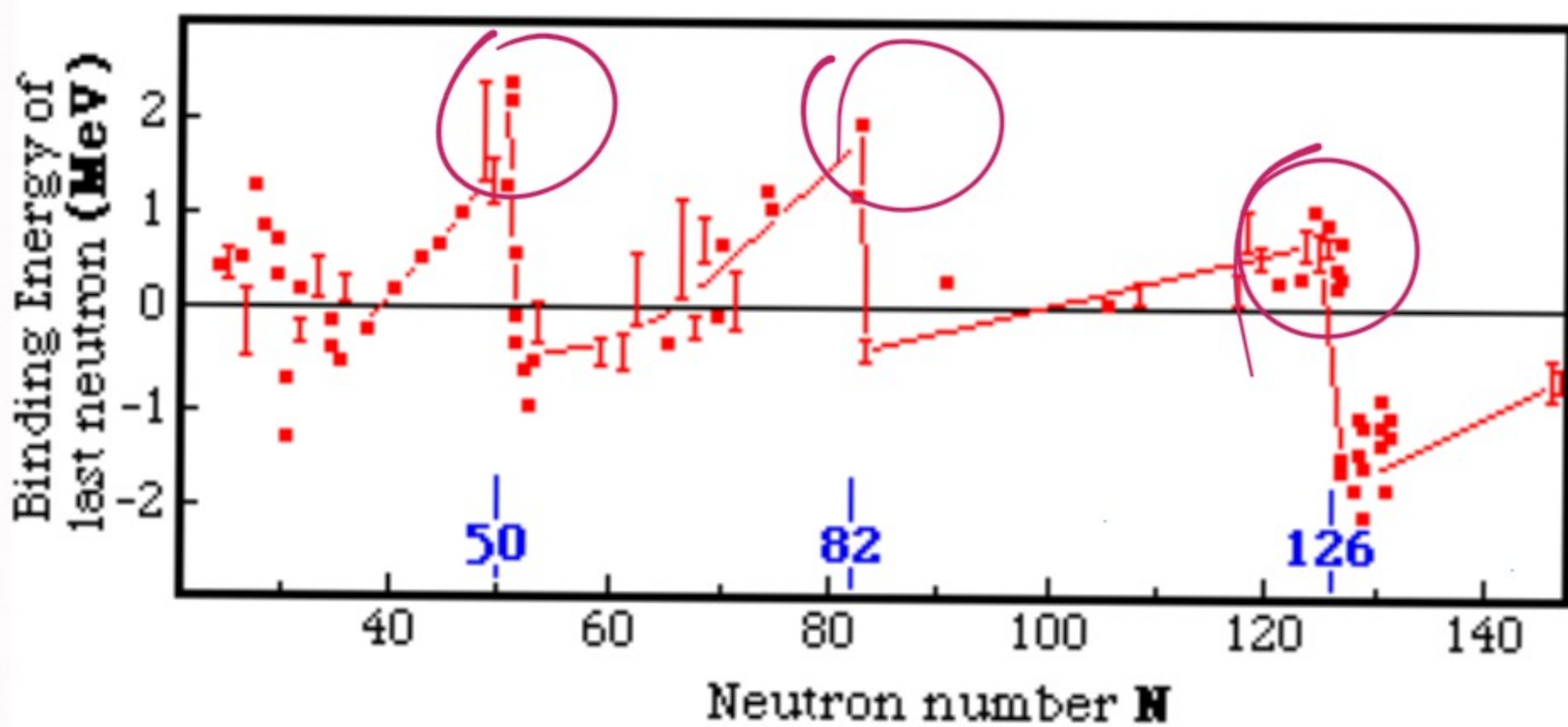
This curve gives us a fusion & fission region

Separation energy \rightarrow another useful quantity

$$S_p(Z, N) = B(Z, N) - B(Z-1, N)$$

$$S_n(Z, N) = B(Z, N) - B(Z, N-1)$$

$\rightarrow \Rightarrow$ something curious here



\rightarrow For some (N, Z) the separation energy reaches a maximum

$N, Z = 2, 8, 20, 28, 50, 82, \dots$
(magic numbers)

Another type of separation energy:

$$S_{\alpha}(N, Z) = B(Z, N) - B(Z-2, N-2) - B(2, 2)$$

α -particle separation energy

$$A > 150 \Rightarrow S_{\alpha} < 0$$

most heavy nuclei can decay by emitting an α -particle

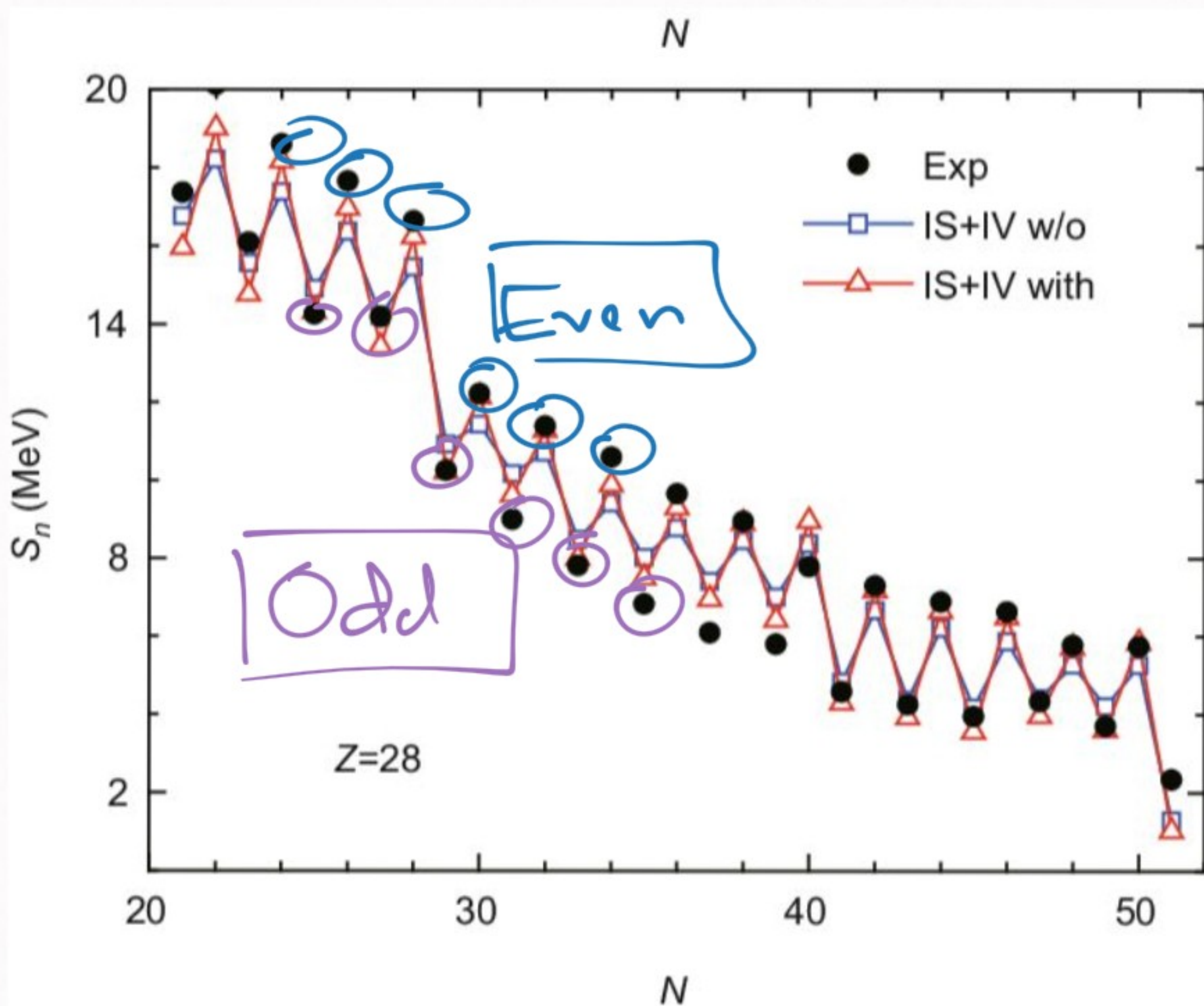
We have also a separation energy related to "pairs"

$$\delta_n = S_n(Z, N) - S_n(Z, N-1)$$

w/ N even, $N-1$ odd

$\delta_n \approx 2\text{MeV}$

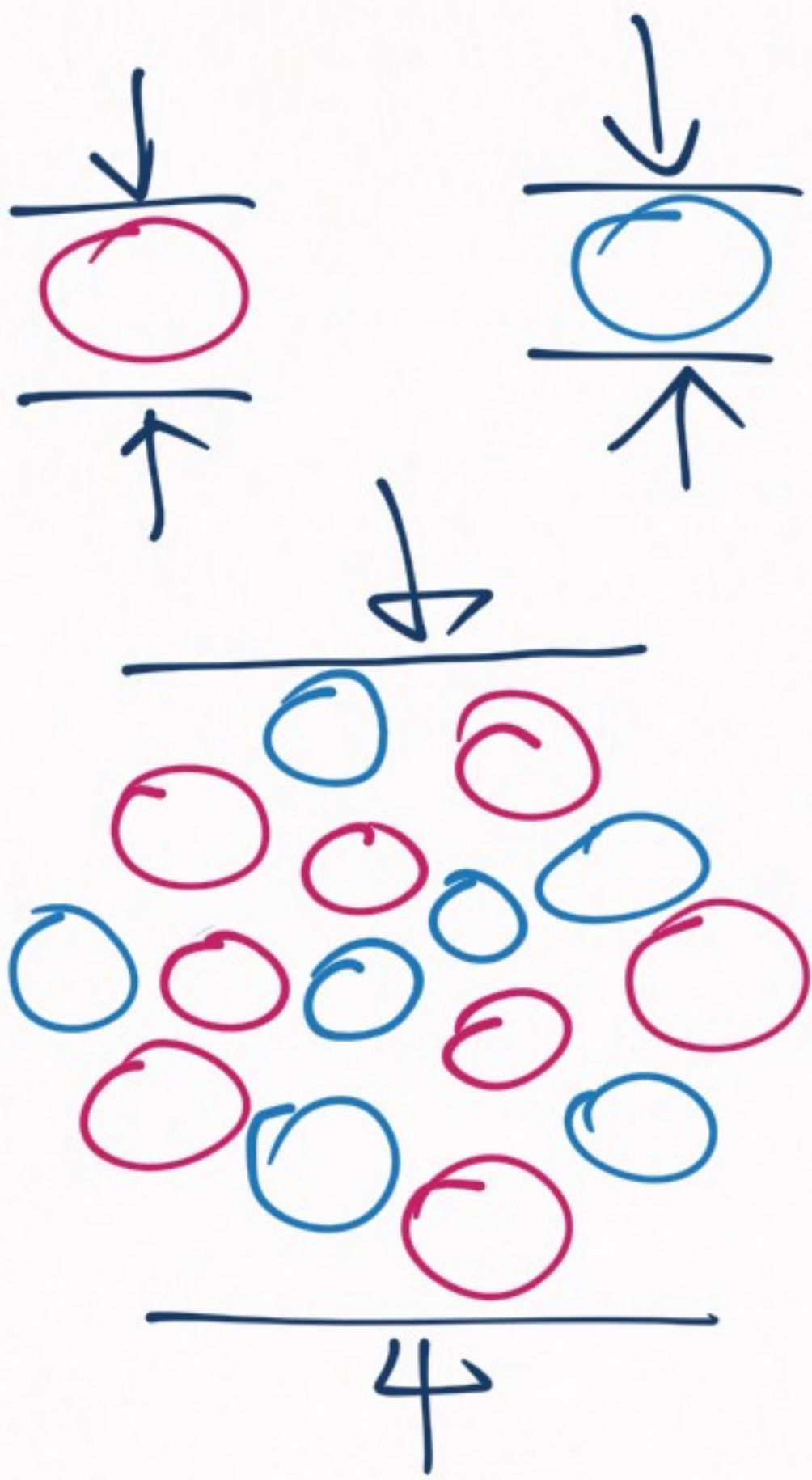
EXAMPLE



→ even # of neutrons / protons
more stable

Nuclear properties:

2) Nuclear size



$$\sqrt{\langle r^2 \rangle_N} \sim (0.5 - 1.0) \text{ fm}$$

$$R^3 \sim A r_0^3$$

$$w / r_0 \geq \sqrt{\langle r^2 \rangle_N}$$

$$r_0 \sim 1.2 - 1.3 \text{ fm}$$

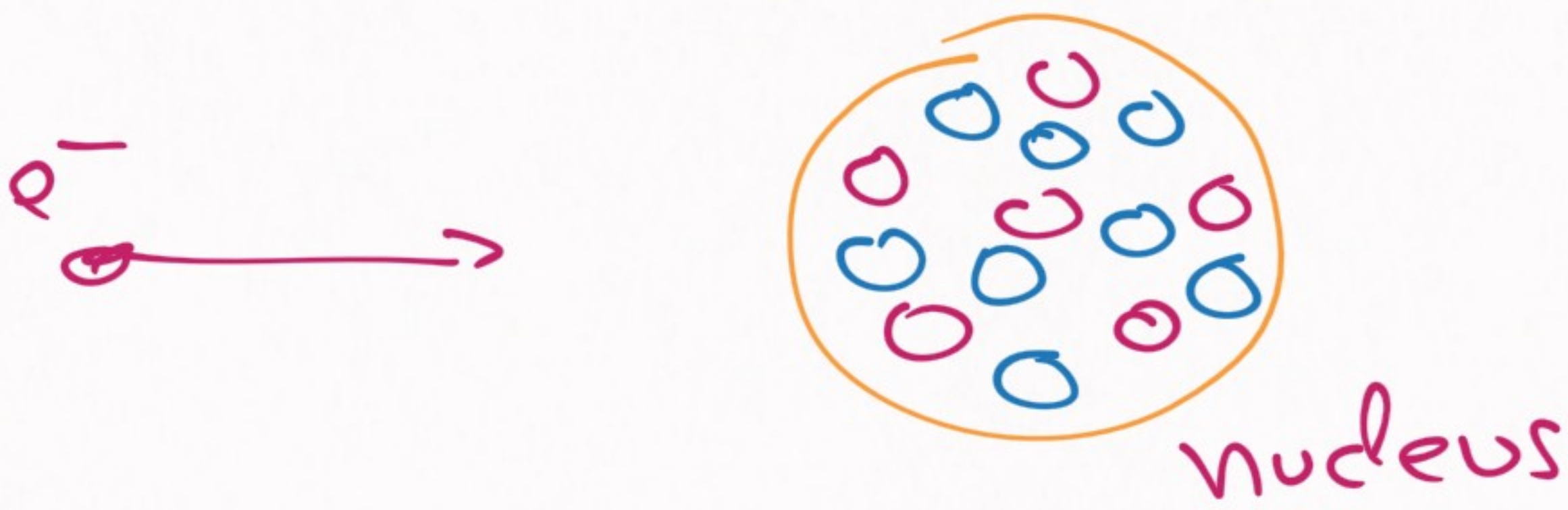


$$R \sim A^{1/3} r_0$$

But there is more ...

The Hofstadter experiment

→ Form factors



$$\frac{d\sigma}{d\Omega} = \left(\frac{d\sigma}{d\Omega} \right)_{\text{pointlike}} |F(\vec{q}^2)|^2$$

$\frac{d\sigma}{d\Omega}$ if nucleus
was a point

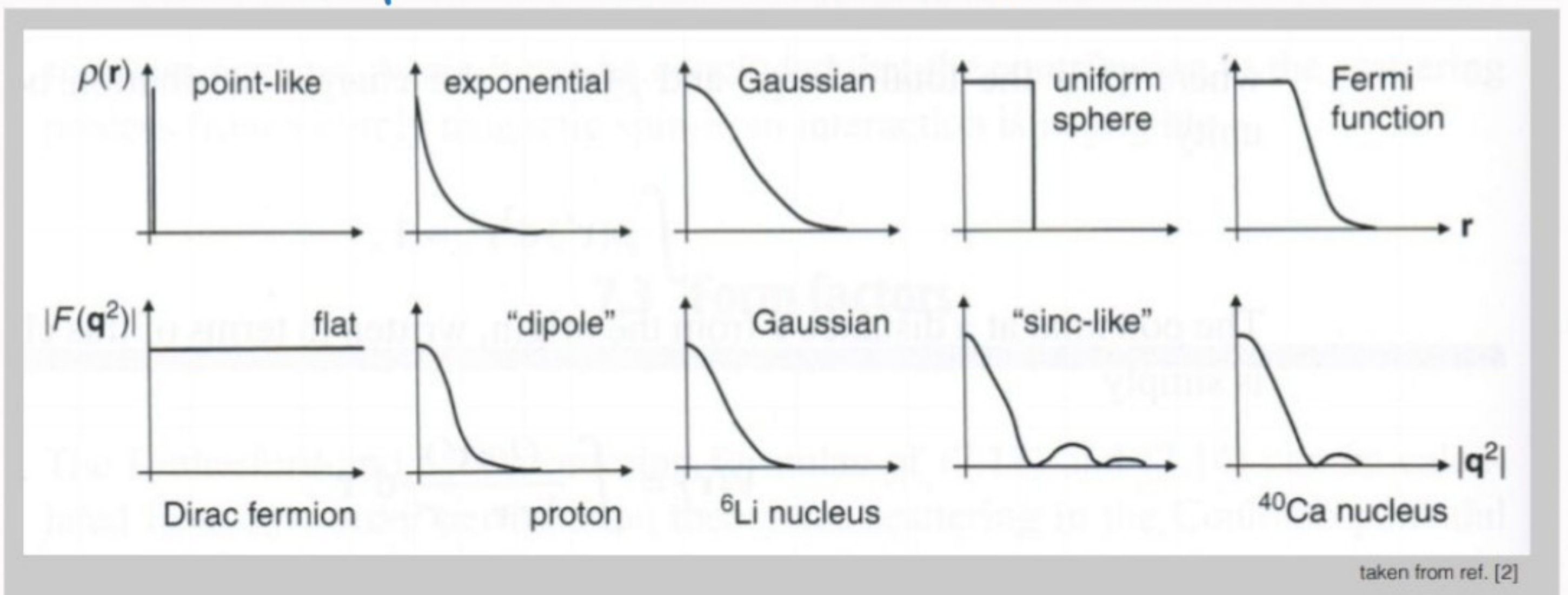
Form
factor

Internal structure of
nucleus

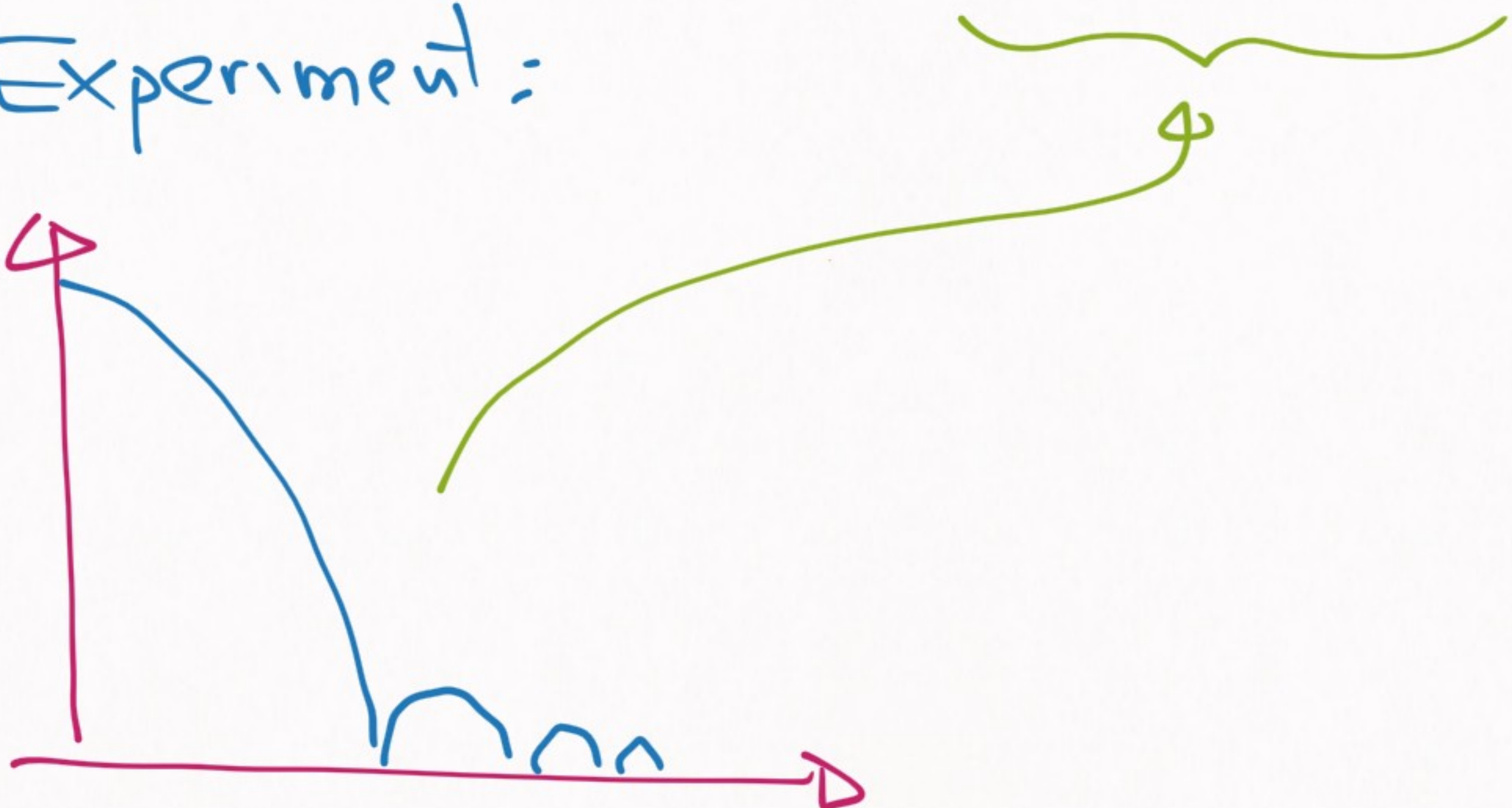
Form factor ρ density:

$$F(\vec{q}) = \int d^3\vec{r} \underbrace{|\psi(\vec{r})|^2}_{\rho(\vec{r})} e^{-i\vec{q}\cdot\vec{r}}$$

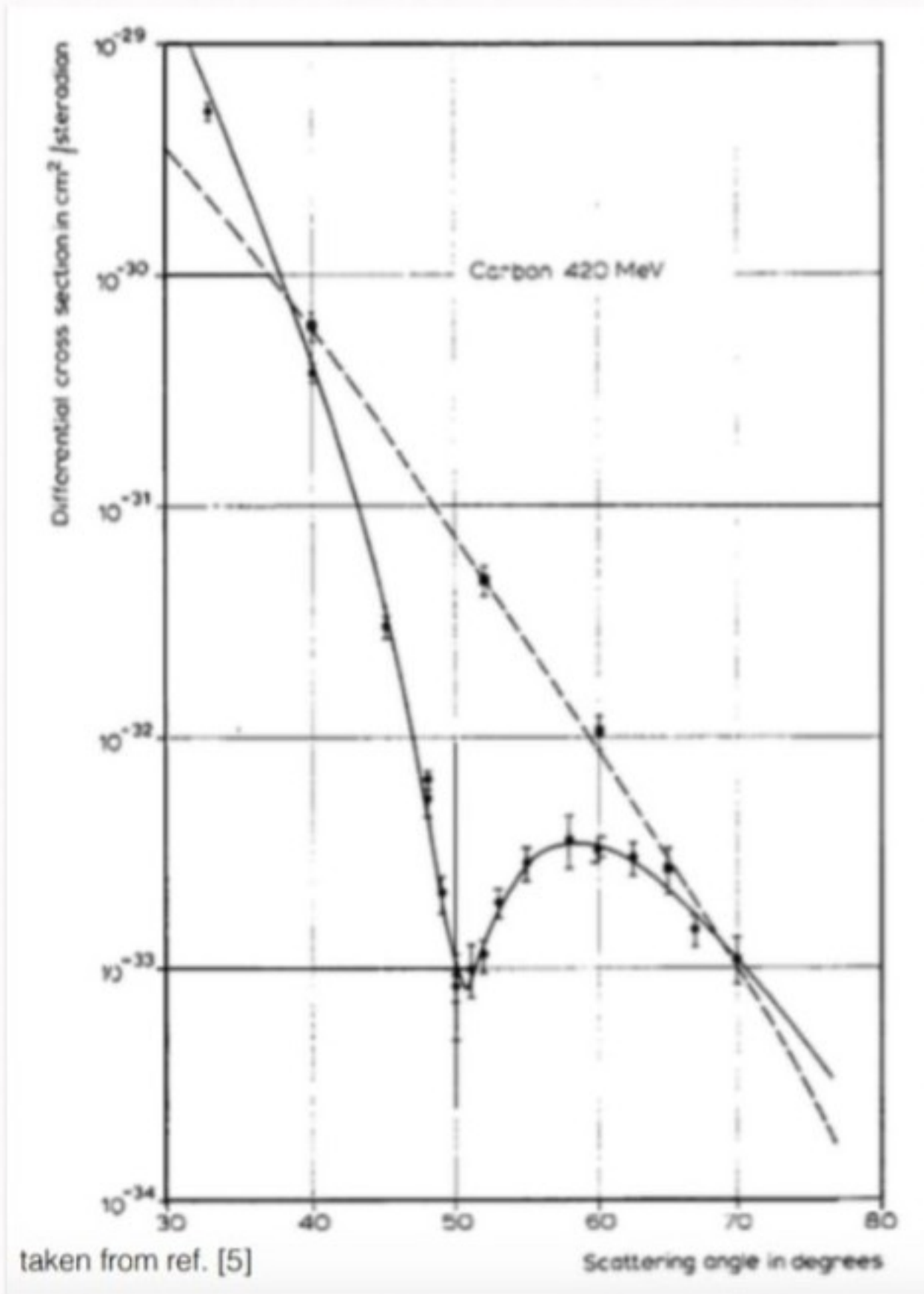
A few possible examples:



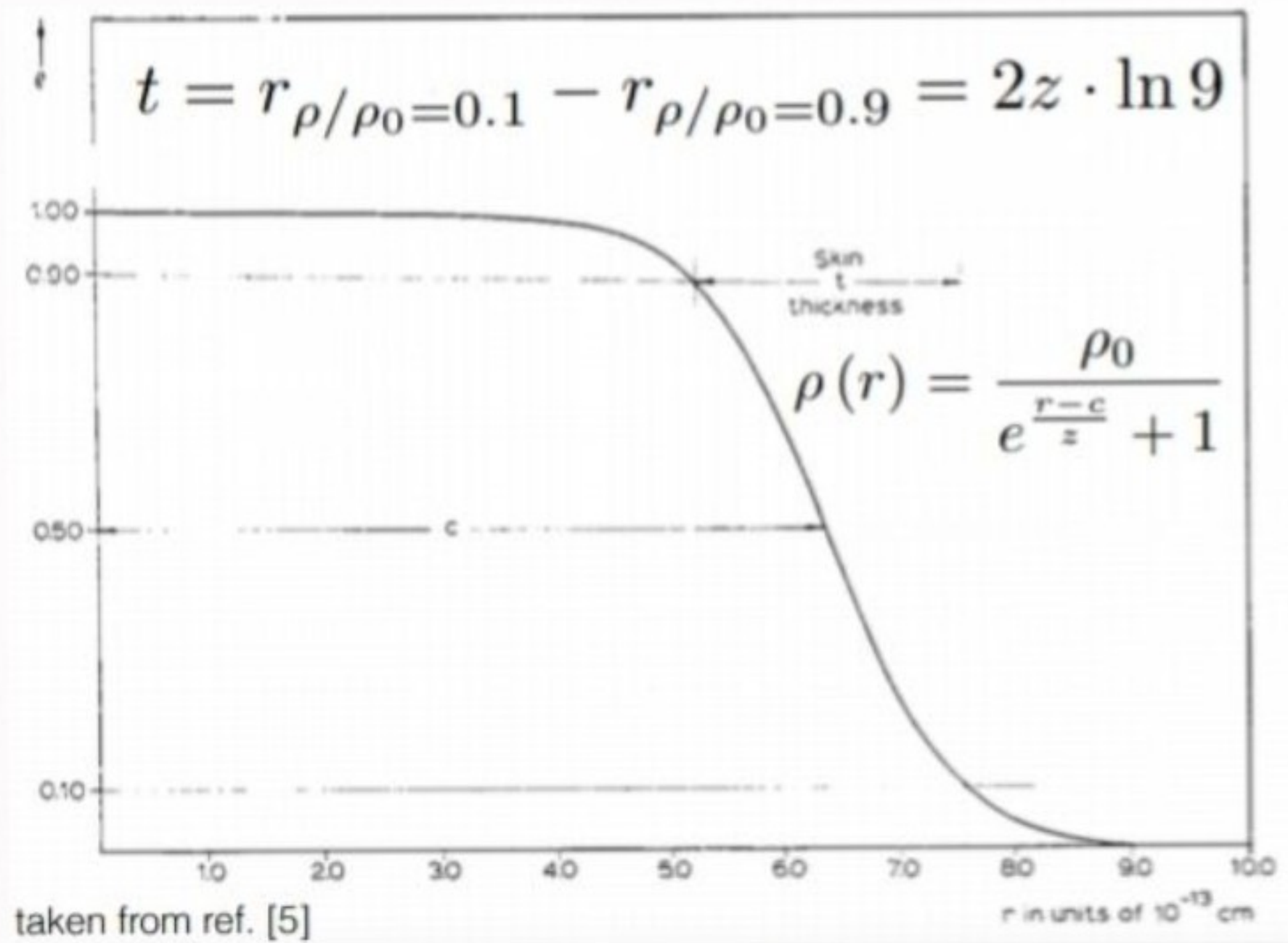
Experiment:



In particular Hofstadter found



From which he deduced



Woods-Saxon distribution

Typical mass distribution:

$$\rho(r) = \frac{\rho_0}{1 + e^{(r-R_0)/a_0}}$$

w/ $\rho_0 \approx 0.17 \text{ fm}^{-3}$

$$a_0 \approx 0.54 \text{ fm}$$

$$R_0 \approx 1.128 A^{1/3} \text{ fm}$$

Nuclear properties:

3) Angular momentum
& Parity

$$J^P(\pi) = 0^- \quad J^P(\rho) = 1^+$$

$$J^P(\sigma) = 0^+$$

$$J^P(N) = \frac{1}{2}^+ \quad N = n, p$$

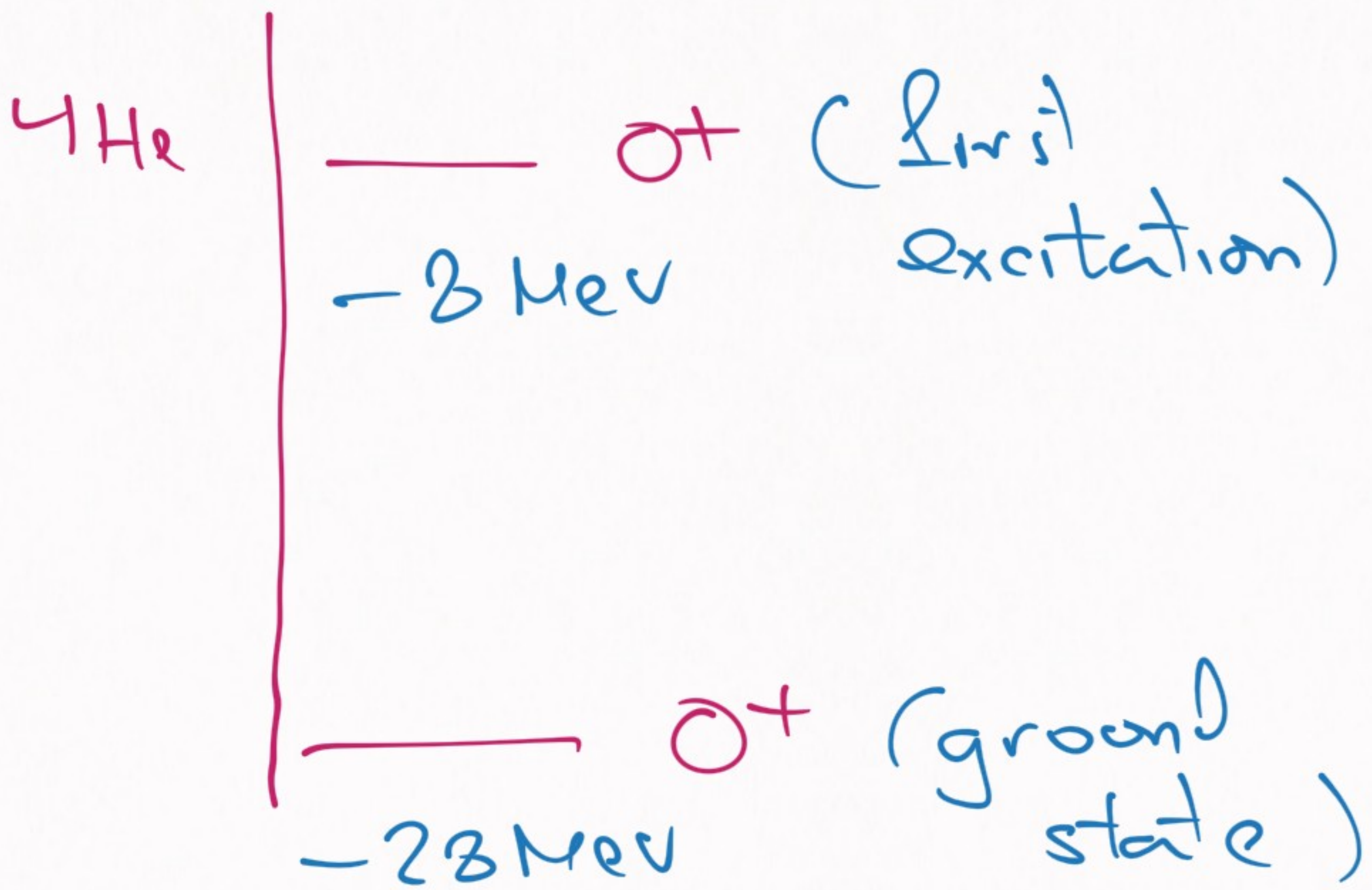
$$J^P(d) = 1^+$$

$$J^P(^3H/^3He) = \frac{1}{2}^+$$

$$J^P(^4He) = 0^+$$

And so on...

Most nuclei have excited states



\Rightarrow Explaining J^P of excited states will be a very important part of nuclear models

Excited states J^π and energy related to nuclear structure:

1) ${}^4\text{He}$ (0^+) (0^+)

↳ "Breathing" mode of collective model

2) ${}^{41}\text{Ca}$ $\begin{array}{l} \text{--- } 3/2^+ \\ \text{--- } 3/2^- \\ \text{--- } 7/2^- \end{array}$ } Typical shell model spectrum

3) ${}^{122}\text{Te}$ $\begin{array}{l} \equiv 0^+, 2^+, 4^+ \\ \text{--- } 2^+ \\ \text{--- } 0^+ \end{array}$ } vibrational spectrum

4) ${}^{166}\text{Er}$ $\begin{array}{l} \text{--- } 6^+ \\ \text{--- } 4^+ \\ \text{--- } 2^+ \\ \text{--- } 0^+ \end{array}$ } rotational spectrum

We will study all this
in the next few lessons



Next lesson:

→ Electromagnetic properties
of nuclei